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react with metals.

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Salts Acid is hydrogen-ion donor ( $H^+$  or proton);

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Acids and Bases.

Types of Arrhenius

Acids. Taste sour,

turns litmus paper

red, conducts

electricity. Tastes

bitter, feels slippery,

turns litmus paper

blue, conduc $\square$ . Acids:

Compounds that yield

hydrogen ions ( $H^+$ ) in

an aqueous s $\square$ .



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An acid is a  
substance that  
contains hydrogen  
and ionizes to  
produce hydrogen  
ions in aqueous  
solution; A base is a

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Substance that  
contains a hydroxide  
group and dissociates  
to produce a  
hydroxide ion in  
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Bronsted-Lowry  
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