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AP ® CHEMISTRY 2010 SCORING GUIDELINES Question 3 (9 points) $8\text{H}^+(\text{aq}) + 4\text{Cl}^-(\text{aq}) + \text{MnO}_4^-(\text{aq}) \rightarrow 2\text{Cl}_2(\text{g}) + \text{Mn}^{3+}(\text{aq}) + 4\text{H}_2\text{O}(\text{l})$ $\text{Cl}_2(\text{g})$ can be generated in the laboratory by reacting potassium permanganate with an acidified solution of sodium chloride. The net-ionic equation for the reaction is given above. (a) A 25.00 mL sample of 0.250 M

ap10 chemistry scoring guidelines final

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AP® CHEMISTRY 2010 SCORING GUIDELINES (Form B) Question 2 (10 points) $5\text{Fe}^{2+}(\text{aq}) + \text{MnO}_4^-(\text{aq}) + 8\text{H}^+(\text{aq}) \rightarrow 5\text{Fe}^{3+}(\text{aq}) + \text{Mn}^{2+}(\text{aq}) + 4\text{H}_2\text{O}(\text{l})$ A galvanic cell and the balanced equation for the spontaneous cell reaction are shown above. The two reduction half-reactions for the overall reaction that occurs in the cell are shown in the table below.

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2010 AP Chemistry Free Response (Form B) Question #1

AP Chemistry Free Response Answers. Adrian has 30 years of high school and early college chemistry teaching experience, in both the UK and the USA.

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